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Key indicators

Single-crystal X-ray study T = 130 KMean σ (C–C) = 0.003 Å R factor = 0.039 wR factor = 0.092 Data-to-parameter ratio = 12.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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exo-7-Phenyl-3-*n*-propyl-5-oxa-2-thia-6azabicyclo[3.2.0^{1,4}]hept-6-ene 2,2-dioxide

The title compound, $C_{13}H_{15}NO_3$, contains the novel *exo*thiabicyclo[3.2.0^{1,4}]hept-6-ene ring system, with pendant phenyl and *n*-propyl substituents. Both fused rings in the bicyclic system are planar, their planes forming a dihedral angle of 65.6 (1)°. Received 14 May 2002 Accepted 30 May 2002 Online 8 June 2002

Comment

The title compound, (I), was prepared as part of a study of the use of carnivore odours in mammal pest control; it is closely related to a previously reported structure of *exo-7-aza-6-oxa-4-n*-propyl-3-thiabicyclo[$5.4.1^{1,7}1^{2,5}$]undecane 3,3-dioxide (Woolhouse *et al.*, 1993), which had been obtained from a thiete sulfone (Gainsford & Woolhouse, 1994).



The crystal structure of (I) is built of isolated molecules (Fig. 1) associated into infinite chains along the *b* axis of the crystal *via* weak intermolecular contacts $C14-H14A\cdots O5^{i}$ [symmetry code: (i) 1/2-x, 1/2+y, 1/2-z], with $H14A\cdots O5$ and $C14\cdots O5$ distances of 2.56 (2) and 3.343 (3) Å, respectively.

The fused four- and five-membered rings are each planar, with average deviations of 0.015 (2) and 0.011 (2) Å, respectively; their least-squares planes form a dihedral angle of 65.6 (1)°. The pendant planar phenyl ring (C8–C13) is twisted by 7.3 (1)° from the five-membered C₃NO ring.

Only two other compounds have been reported [Allen & Kennard (1993) and *ConQuest* (Cambridge Crystallographic Data Centre, 2002)] with an oxygen bound to the C_3SO_2 ring. Each of these (Beagley *et al.*, 1992; Adiwidjala *et al.*, 2000) have the oxygen bound to the carbon remote from the sulfur heteroatom, as is found when the fused ring system is constructed by cycloaddition.

Experimental

To an ethereal solution of 2-*n*-propyl- Δ^3 -thiete sulfone (0.68 g, 4.7 mmol) and benzhydroxamoyl chloride (0.8 g, 5.1 mmol) at 273 K was added, dropwise, a solution of triethylamine (0.52 g, 5.1 mmol) in ether. The solution was stirred for 4 h at ambient temperature, then filtered and concentrated. The compound was obtained with the *endo*-fused stereoisomer (0.62 g, 50%) by flash chromatography over silica (0.42 g, 34% yield); crystals were grown from an ethylacetate/ hexane mixture.

organic papers

Crystal data

 $C_{13}H_{15}NO_3S$ $M_r = 265.32$ Monoclinic, $P2_1/n$ a = 12.593 (5) Å b = 5.081 (2) Å c = 20.169 (8) Å $\beta = 104.028$ (15)° V = 1252.0 (9) Å³ Z = 4

Data collection

Siemens/Nicolet *R3m* four-circle diffractometer ω scans Absorption correction: none 2933 measured reflections 2794 independent reflections 1801 reflections with $I > 2\sigma(I)$ $R_{int} = 0.036$

Refinement

 Refinement on F^2 All H-atom parameters refined

 $R[F^2 > 2\sigma(F^2)] = 0.039$ $w = 1/[\sigma^2(F_o^2) + (0.0523P)^2]$
 $wR(F^2) = 0.092$ where $P = (F_o^2 + 2F_c^2)/3$

 S = 0.85 $(\Delta/\sigma)_{max} < 0.001$

 2794 reflections
 $\Delta\rho_{max} = 0.29$ e Å⁻³

 223 parameters
 $\Delta\rho_{min} = -0.69$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

S2-O21	1.4451 (17)	O5-C4	1.453 (2)
S2-O22	1.4413 (15)	N6-C7	1.298 (2)
S2-C1	1.834 (2)	C1-C7	1.502 (3)
S2-C3	1.835 (2)	C1-C4	1.545 (3)
O5-N6	1.411 (2)	C3-C4	1.557 (3)
022 82 021	118 61 (0)	C4 C1 S2	80.86 (13)
$C_{1}^{-S_{2}^{-}-C_{3}^{-}}$	80.68 (10)	$C_{4} - C_{1} - S_{2}$	89.46 (13)
N6-O5-C4	109.94 (14)	O5-C4-C1	104.82 (17)
C7-N6-O5	110.41 (16)	O5-C4-C3	113.08 (17)
C7-C1-C4	101.85 (16)	C1-C4-C3	99.92 (16)
C7-C1-S2	112.64 (15)	N6-C7-C1	112.90 (18)
C4-O5-N6-C7	-2.9 (2)	O5-N6-C7-C8	177.53 (17)
S2-C1-C4-O5	-114.97 (14)	S2-C1-C7-N6	95.10 (19)
C7-C1-C4-C3	115.42 (17)	S2-C1-C7-C8	-80.6 (2)

 $D_x = 1.408 \text{ Mg m}^{-3}$

Cell parameters from 24

Mo $K\alpha$ radiation

reflections

 $\theta = 5.6 - 14.2^{\circ}$ $\mu = 0.26 \text{ mm}^{-1}$

T = 130 (2) K

 $\theta_{\rm max} = 27.9^{\circ}$

 $\begin{array}{l} h=0 \rightarrow 16 \\ k=0 \rightarrow 6 \end{array}$

 $l=-26\rightarrow 25$

3 standard reflections

every 97 reflections

intensity decay: none

Needle, colourless

0.44 \times 0.19 \times 0.04 mm

All H atoms were refined with isotropic displacement parameters. The C-H bonds are in the range 0.92 (3)–1.01 (3) Å.

Figure 1

The molecular structure of (I) (Farrugia, 1997). Displacement ellipsoids are drawn at the 50% probability level. H atoms have arbitrary radii.

Data collection: R3M Software (Siemens, 1983); cell refinement: R3M Software; data reduction: R3M Software; program(s) used to solve structure: SHELXS86 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 in WinGX (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 1990).

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References

- Adiwidjala, G., Olbrich, F., Schulze, O. & Voss, J. (2000). Thesis, University of Hamburg, Germany.
- Allen, F. H. & Kennard, O. (1993). Chem. Des. Autom. News, 8, 1, 31-37.
- Beagley, B., James, M. R., Pritchard, R. G., Raynor, C. M., Smith, C. & Stoodley, R. J. (1992). J. Chem. Soc. Perkin Trans 1, pp. 2371–2382.

Cambridge Crystallographic Data Centre (2002). *ConQuest.* Version 1.3. Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, England.

- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Gainsford, G. J. & Woolhouse, A. D. (1994). Acta Cryst. C50, 606-607.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1983). *R3M Software*. Version 4.11. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Spek, A. L. (1990). Acta Cryst. A46, C-34.

Woolhouse, A. D., Gainsford, G. J. & Crump, D. R. (1993). J. Heterocycl. Chem. 30, 873–880.